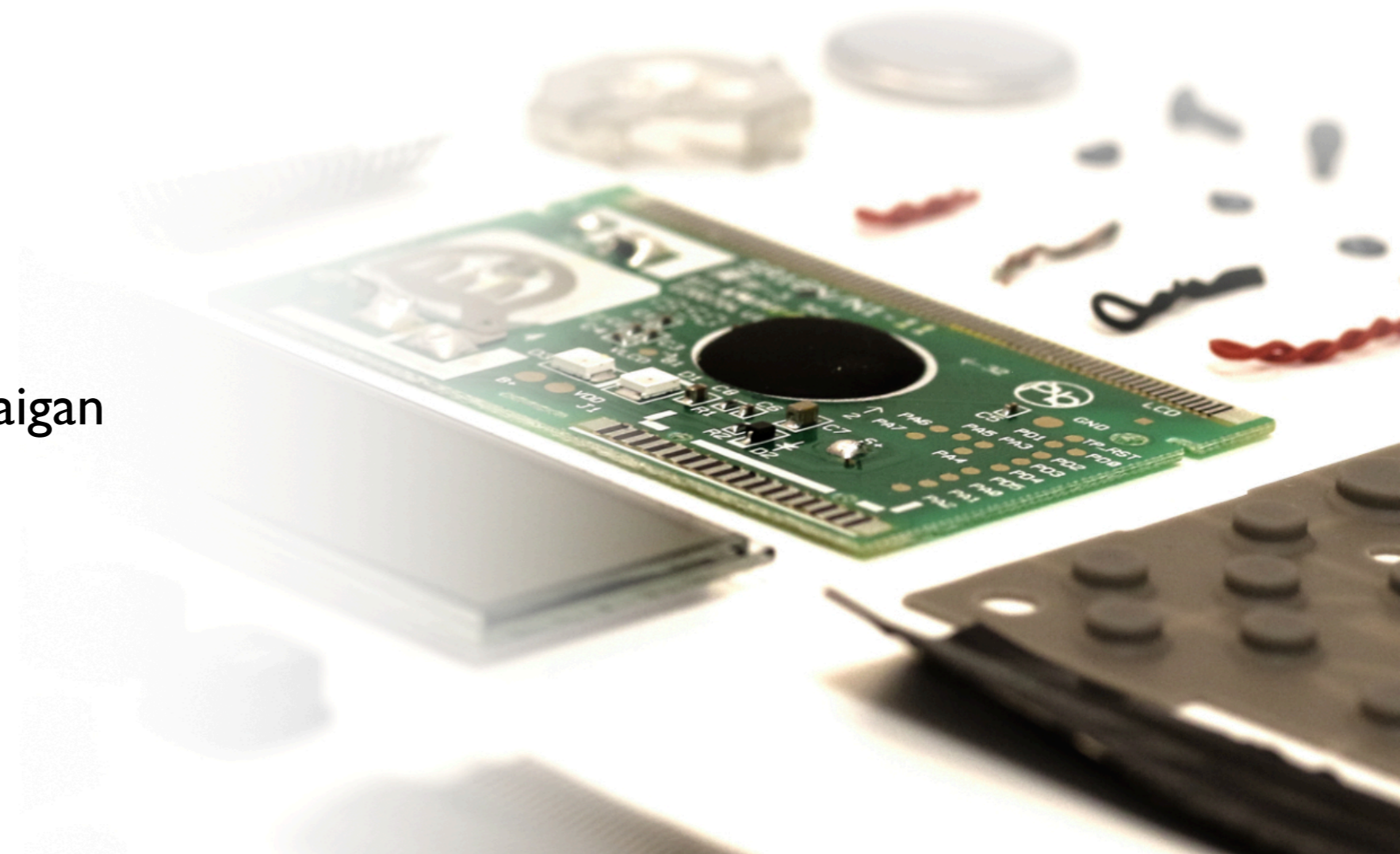


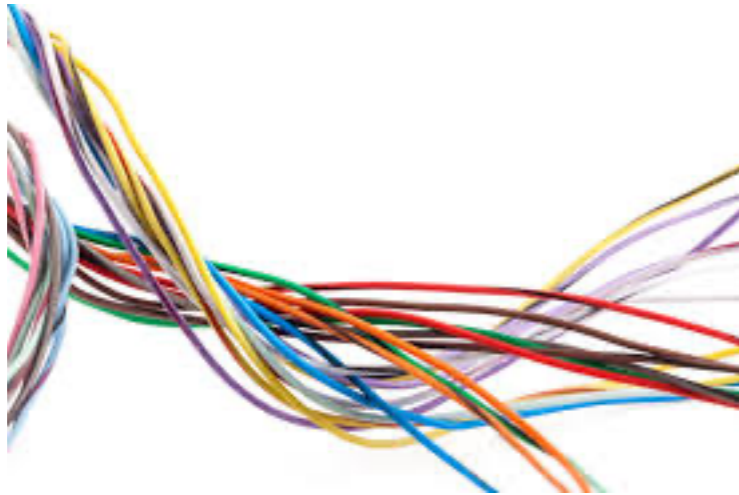
Claigan Webinar PFAS Compliance

Presented by:
Bruce Calder
VP Consulting Services at Claigan

December 14 2022



Overview - Agenda



- What are PFAS?
- Why do we care?
- Why so many PFAS
 - Non-polymer
 - Polymer
- Regulations
 - Reporting
 - Restriction
- Sources of PFAS
- How to comply
- Testing options
- Q&A



PFAS

“Forever Chemicals”



Print subscriptions Search jobs Sign in Search International edition

The Guardian
For 200 years

Home Sport Culture Lifestyle More

Quirks & Quarks

'Forever chemicals' can have far-reaching consequences, need more regulation in Canada, scientists say



'Very significant legislative gap' for PFAS substances, says Environmental Defence's Muhannad Malas

This article is more than 3 months old

Lethal 'forever chemicals' taint our food, water and even blood. The EPA is stalling

David Bond



< Home



HEALTH | News

U.S. EPA unveils strategy to regulate toxic 'forever chemicals'

euobserver



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NEWS / CLIMATE CHANGE

Europe readies to restrict use of toxic 'forever chemicals'

WYPR News

Working to ban a 'forever chemical'

WYPR - 88.1 FM Baltimore | By Joel McCord

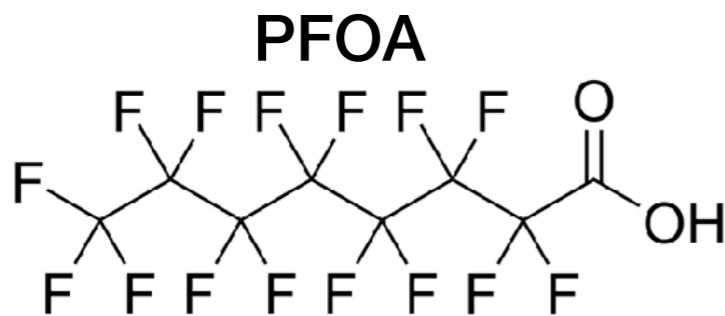
Published February 1, 2022 at 9:58 AM EST



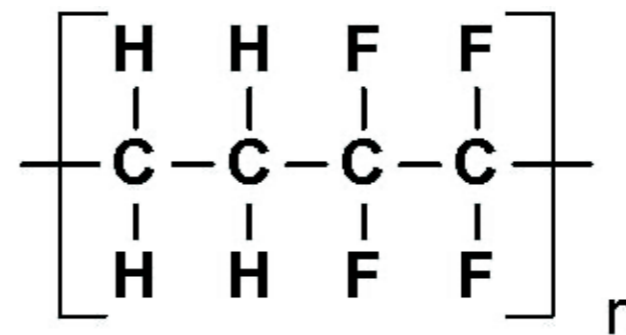
▶ LISTEN • 4:06

What are PFAS

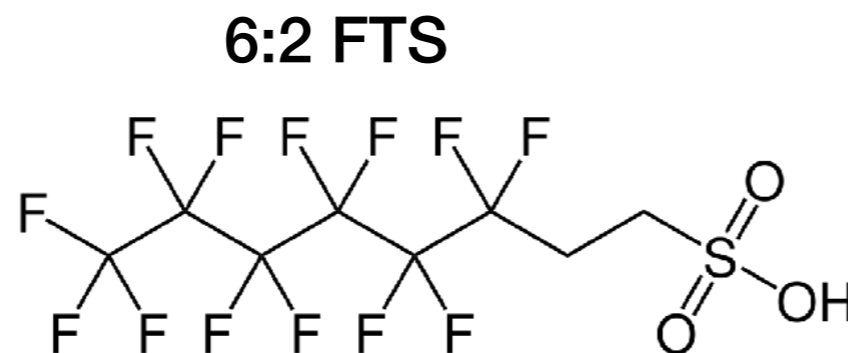
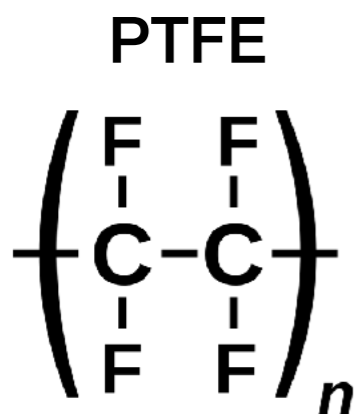
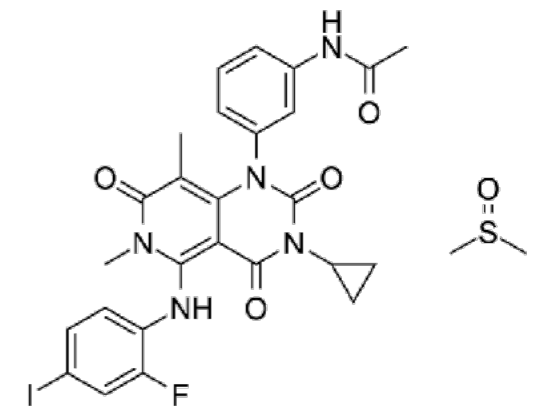
- PFAS (Maine)
 - PFAS (perfluoroalkyl and polyfluoroalkyl substances) means substances that include any member of the class of fluorinated organic chemicals containing one fully fluorinated carbon atom.
- Which means
 - At least one carbon with all hydrogens replaced by fluorines



Fluorinated Polyethylene

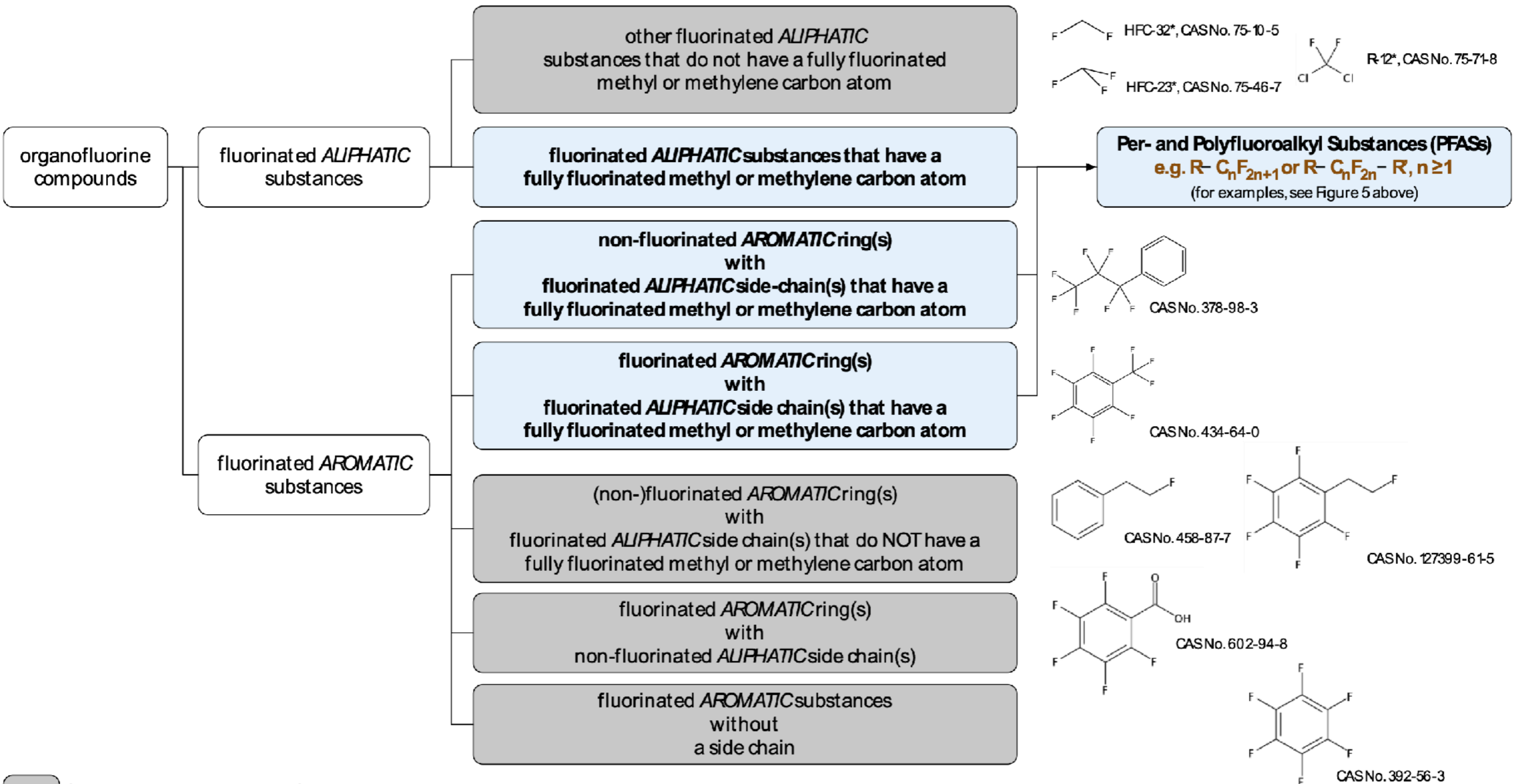


Trametinib



Why so many PFASs?

- Reference



Substances that are not PFASs and are not addressed in this report

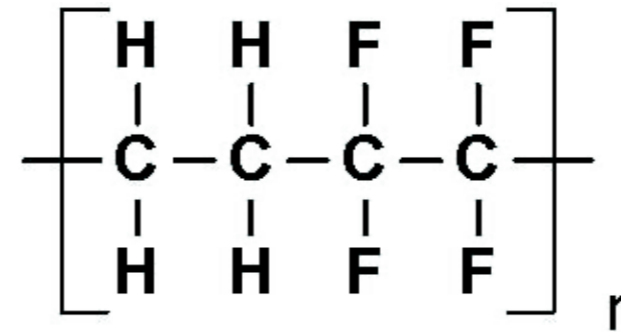
* HFC-32, HFC-23 and R-12 are not PFASs, despite the presence of moieties such as $-CF_2-$ or $-CF_3$, because not all H on the fluorinated carbon atom are replaced by F, i.e., they do not have a fully fluorinated carbon atom.

PFAS

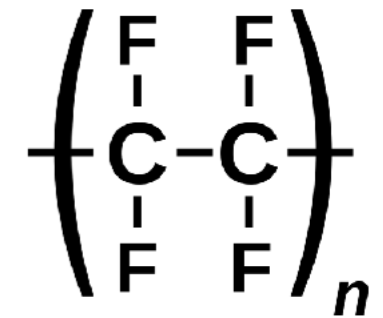
Simplified

- Polymer PFAS
 - PFAS with repeating chains
 - PTFE, PVDF, FKM

Fluorinated Polyethylene

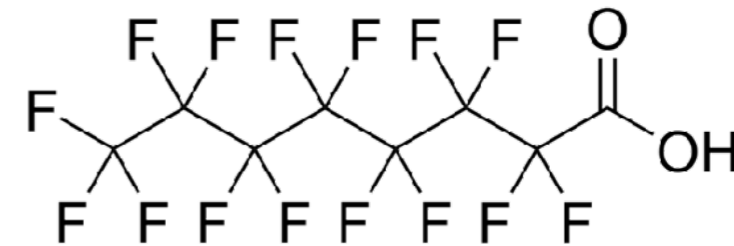


PTFE

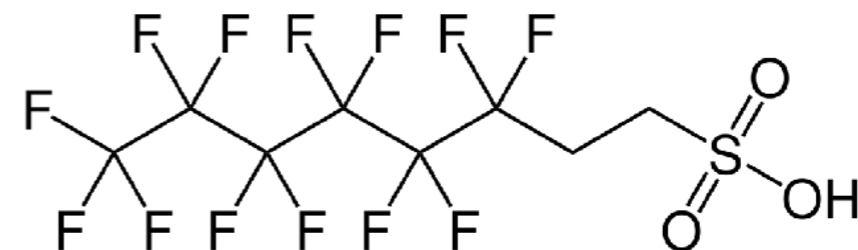


- Non-Polymer (monomer) PFAS
 - PFAS without a repeating chain
 - PFOA, 6:2 FTS, PFOS

PFOA



6:2 FTS



Full Nomenclatures

- If you want a headache

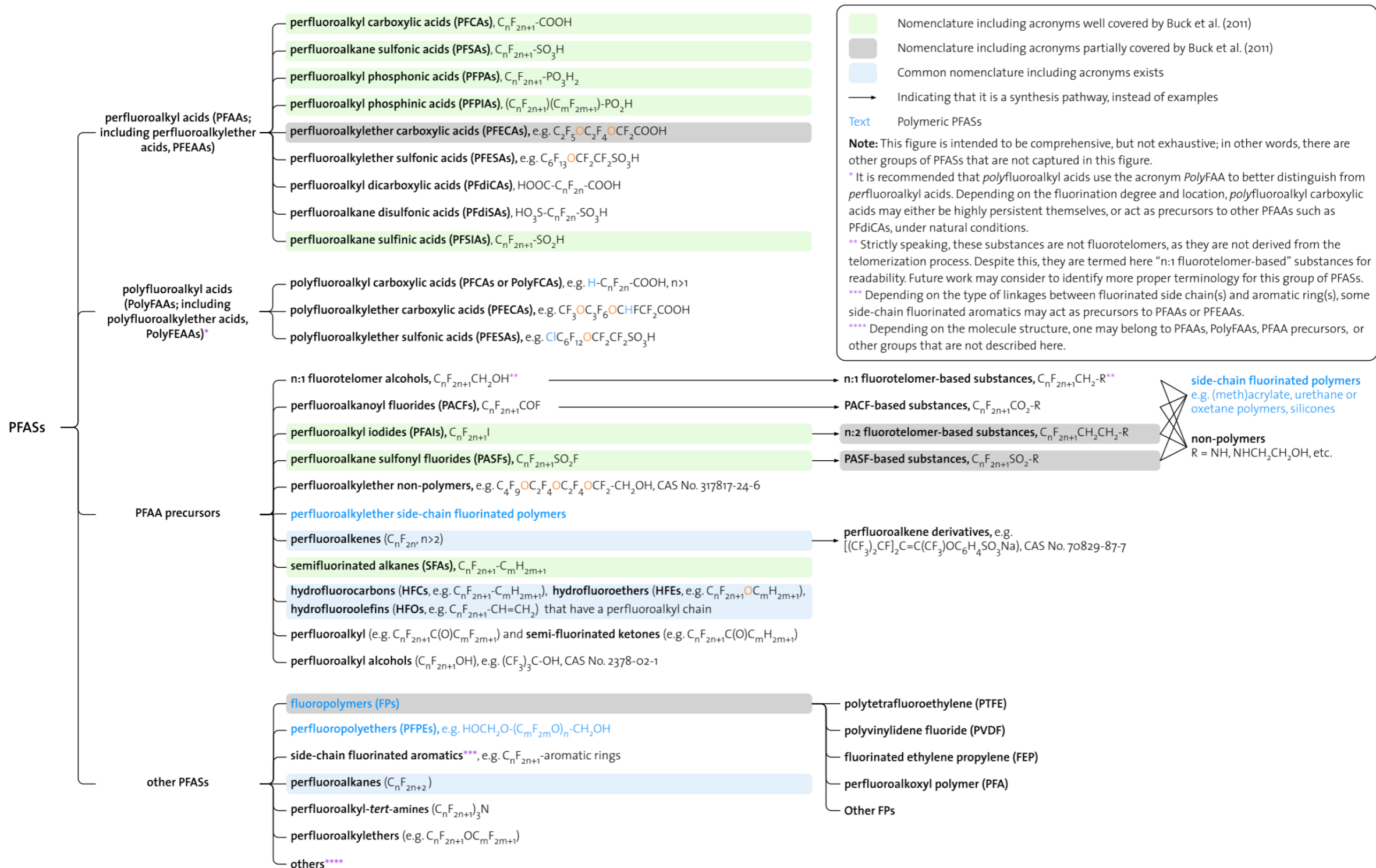


Figure 9. A comprehensive overview of PFAS groups, their structural traits, examples and notes on whether corresponding common nomenclatures (including acronyms)

Why do we care?

- Restricted (PFOA & longer)
 - EU (+related non-EU related countries. ie. Switzerland)
 - California (Proposition 65 and AB1200)
- Planned Restrictions (PFOA & longer)
 - Canada (Canadian Prohibition) - Date TBD
 - Vietnam - Date TBD
- Reporting (Upcoming)
 - Maine
 - US (federally) - TSCA (does not apply to FDA regulated devices)
 - US (state level - may be in parallel to Maine)
 - EU



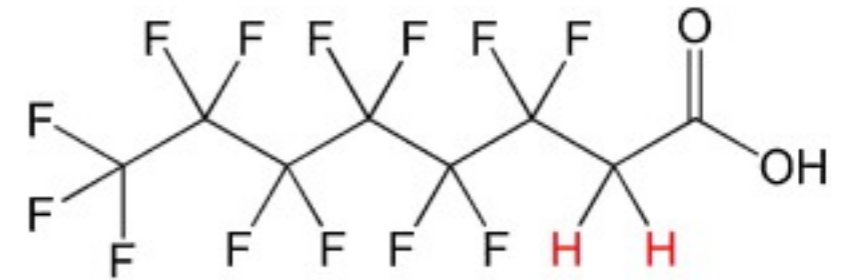
Portal on Per and Poly Fluorinated Chemicals

Why so many PFAS?

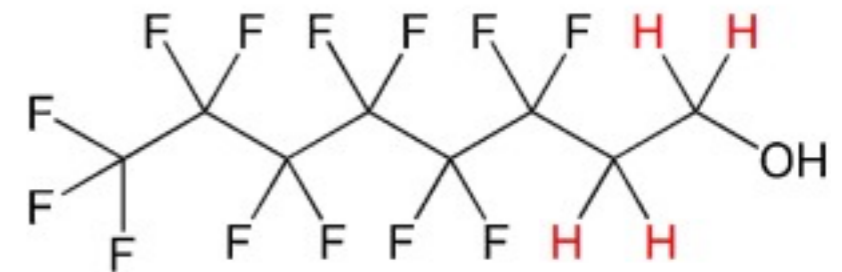
- There are a lot of PFAS substances listed in most regulations
- Why?
 - Most regulations need to list a CAS or EC number
 - The definition includes any chemical with a fully substituted carbon attached to a fluorine
 - Which is a lot
 - 20,000+ potential CAS numbers
- And history has proven that industry can get inventive with fluoroalkyl regulation

Non-Polymer PFAS

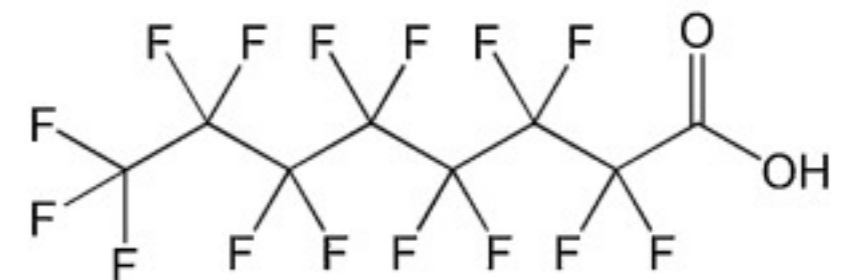
- **Non-Polymer PFAS**
 - Non-repeating PFAS (monomers)
 - Examples
 - PFOA (carboxylate family)
 - PFOS (sulphonate family)
 - Fluorotelomers
 - Fluorotelomer
 - Fluoroacrylates
 - Fluorotelomer sulphides



6:2 FTCA

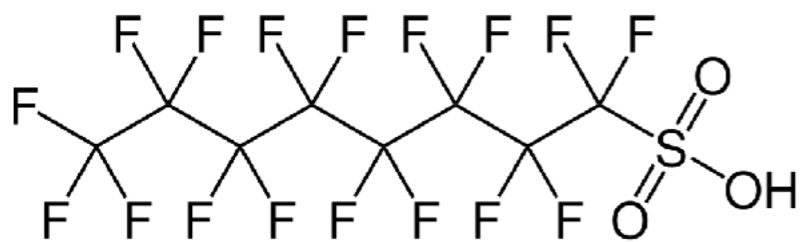


6:2 FTOH

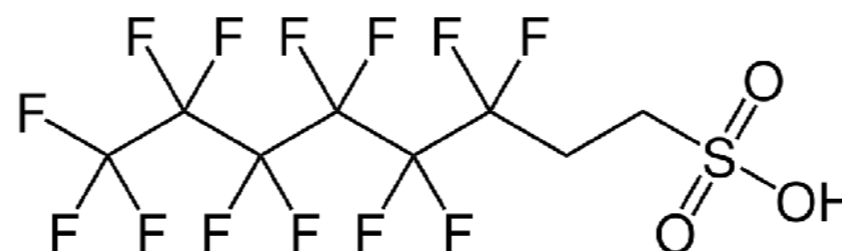


PFOA

PFOS

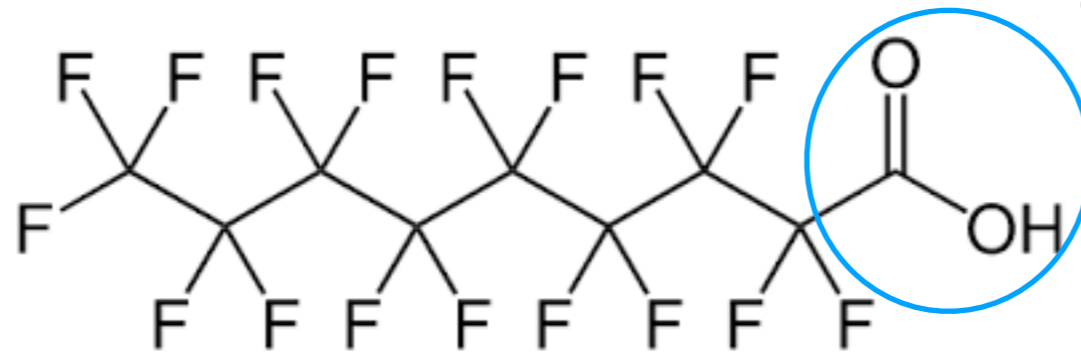


6:2 FTS



Carboxylates

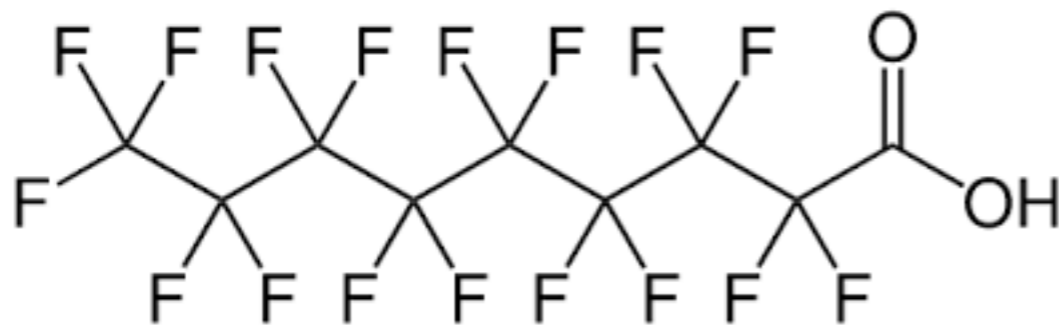
REACH, POP, US TSCA, Prop 65



Carboxylate Group

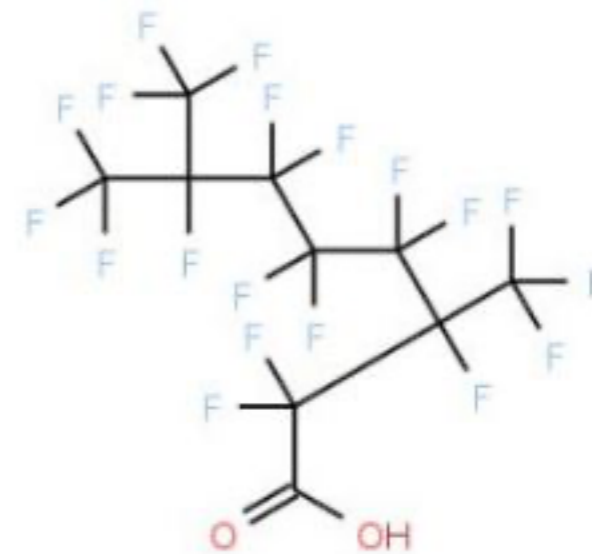
Note - 'H' can be Na, K, NH₄, CH₃ etc..

Linear



PFNA

Branched

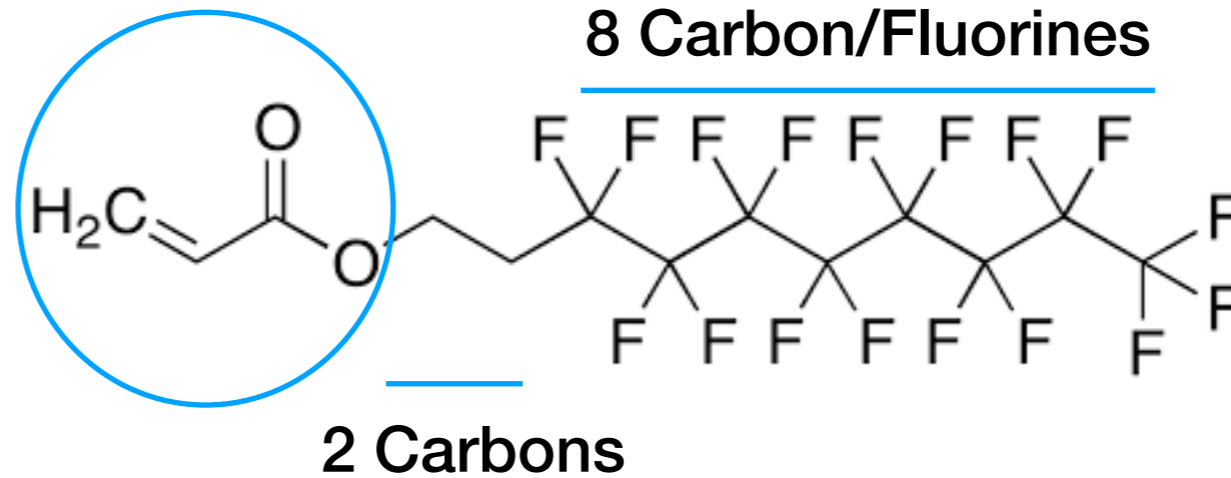


PF-3,7 DMOA

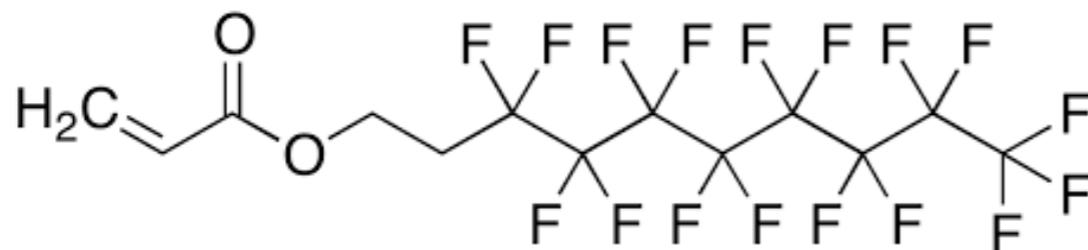
Fluoro-acrylates

REACH Restricted as LCPFAC

Acrylate Group



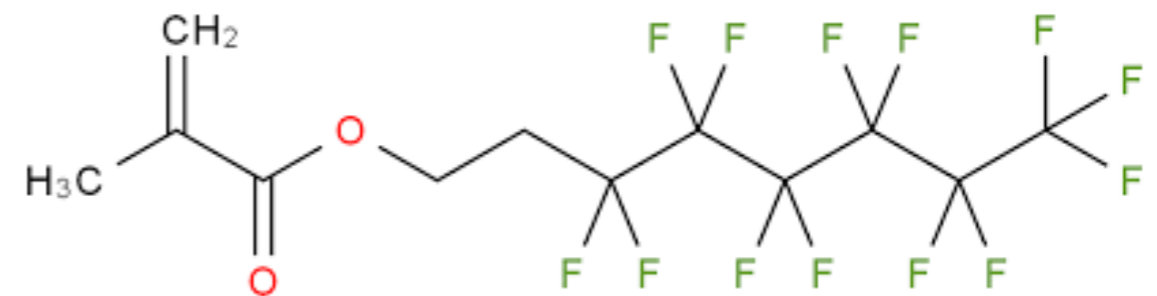
Acrylate



Acrylate

8:2 FTA

Methyl Acrylate



Methyl Acrylate

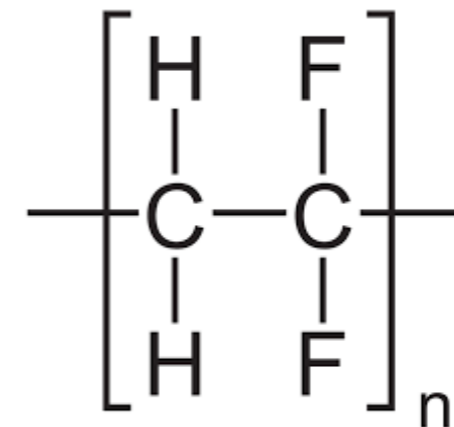
8:2 FTMA

Polymer PFAS Substances

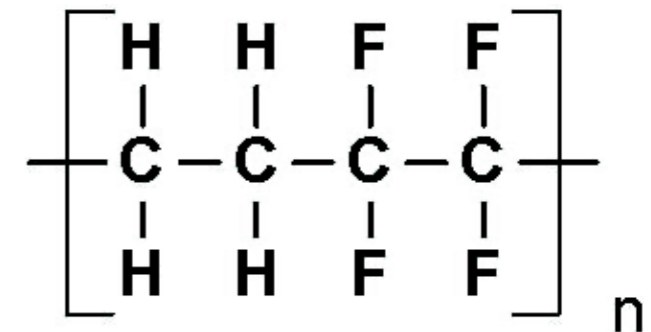
- **Polymer PFAS substances**

- Repeating PFAS (polymers)
- Examples
 - PTFE
 - Fluorinated polyethylene
 - PVDF
 - FKM
 - Fluoroacrylic polymers
 - Fluorosilicone rubber

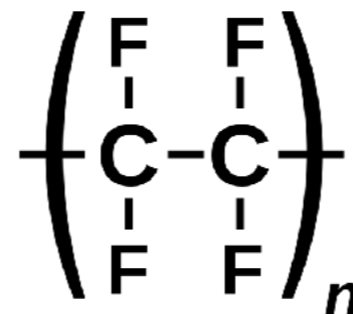
PVDF



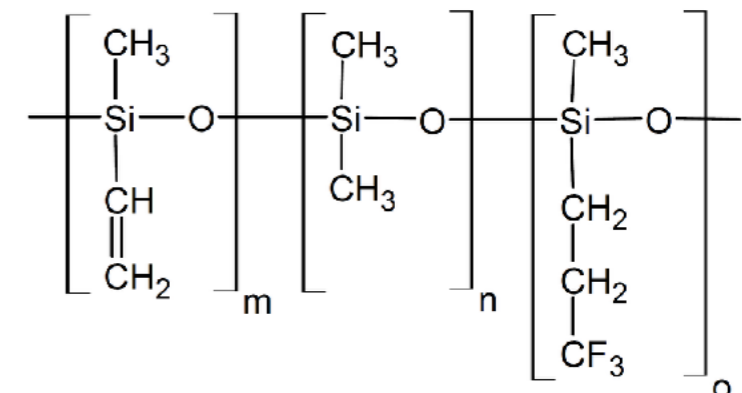
Fluorinated Polyethylene



PTFE

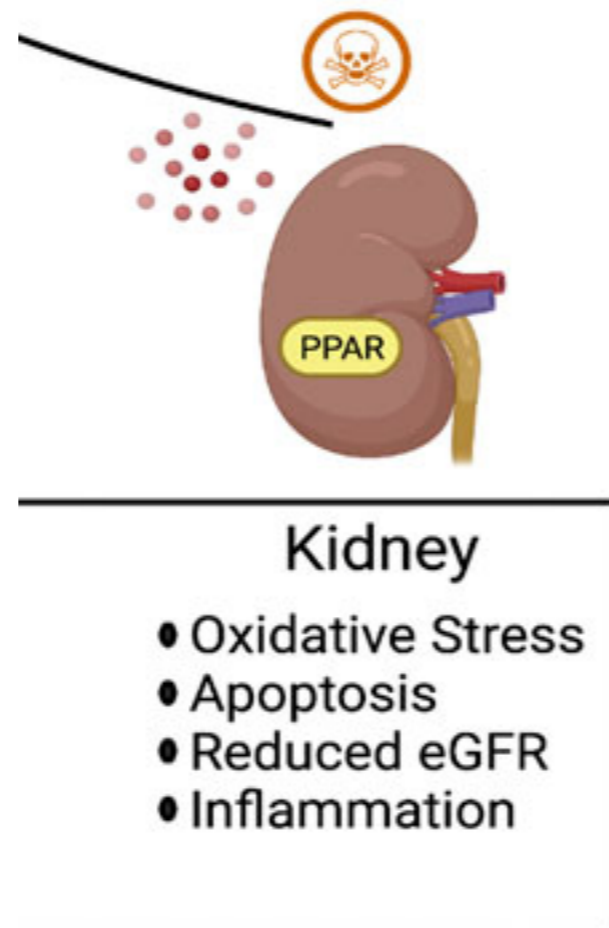


Fluorovinylmethylsiloxane Rubber (FVMQ)



What makes PFASs dangerous?

- **Non-Polymer PFAS**
 - PFOA and similar are surfactants (detergents)
 - Your kidneys use protein surfactants to purge dead cells and fight infection (in particular bacteria).
 - In high concentration, they change the surfactant levels in your kidneys
 - Which affects kidney function and can lead to cancer.



Non-Polymer PFAS Dangers

- **Non-Polymer PFAS** (PFOA and similar)
 - Very little danger at the < 1 ppm level in plastics
 - In comparison
 - Average concentration of PFOA in PTFE -
 - 100 ppb
 - Average concentration of surfactant in laundry detergent -
 - 200,000,000 ppb
 - 2 million times concentration different for same effect
- The most significant different is that fluoroalkyl substances do not degrade in the environmental
- Which makes them consistently measurable in water by highly sensitive equipment (often in the parts per trillion)



What makes PFASs dangerous?

- **Polymer PFAS**

- Very inert with little reactivity with humans or their environment
- If you burn them (>550C / 930F) they release carbon monoxide and a variety of fluorinated substances
- The fluorinated substances are highly toxic but is usually produced in low concentration
- Carbon monoxide is toxic and is usually produced in high concentration



Regulation

Two Types

- **Restriction** (PFOA and similar monomer substances)
 - Restriction on the PFOA substances
 - Such as PFOA and other longer chains perfluoroalkyl carboxylates (LC-PFAC)
 - Or substances that degrade into these substances
 - Example - fluorotelamers and fluoroacrylates



Persistent Organic Pollutants (POPs)

Regulation

Two Types

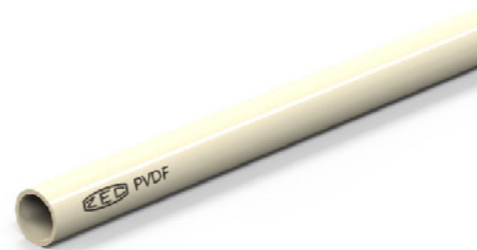
- **Reporting (PFAS)**
 - Reporting intentionally added PFAS
 - Which are generally polymers and polymer coatings
 - Non-polymer substances (PFOA) are **generally not** reportable as they are either
 - A manufacturing aid, or
 - An unintentional byproduct of manufacturing
 - Degradation products can be reportable but are generally covered by the main reportable. Guidance TBD



Polyfluoro uses Solid Polymer

- Solid fluoropolymer made from fluoropolymer

- PTFE connector
- PVDF tube
- FKM o-ring



- **PFOA risk** - as a surfactant in the manufacturing process

- Fluorination of another polymer

- Fluorine and oxygen gas process
- Example - fluorinated polyethylene

- **PFOA risk** - oxygenation generates carboxylates including PFOA and family



Perfluoroalkyl Substances

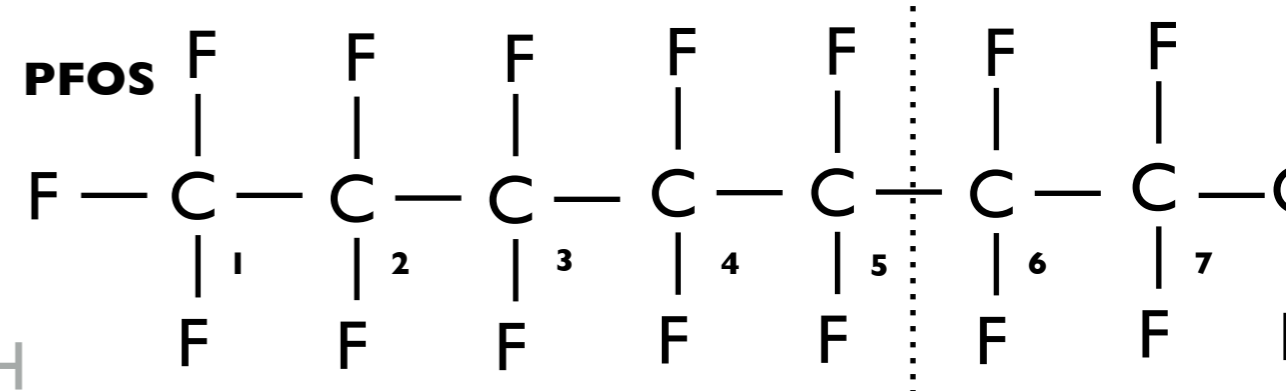
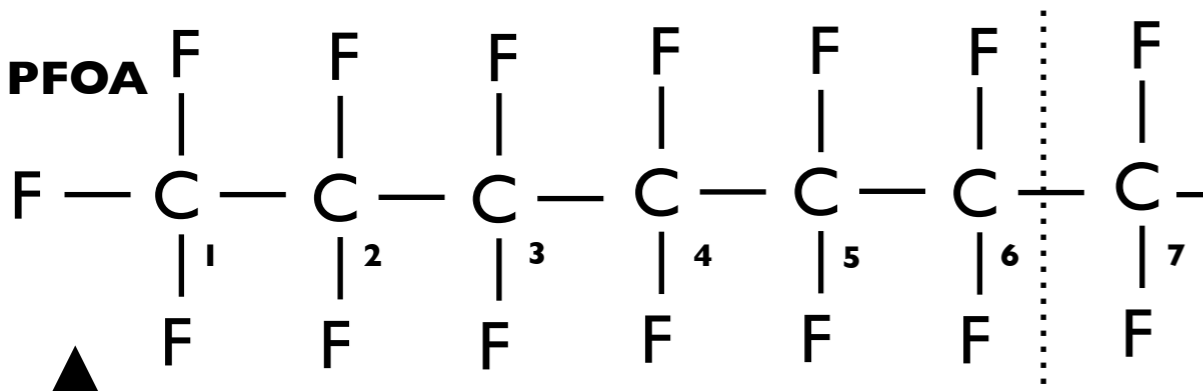
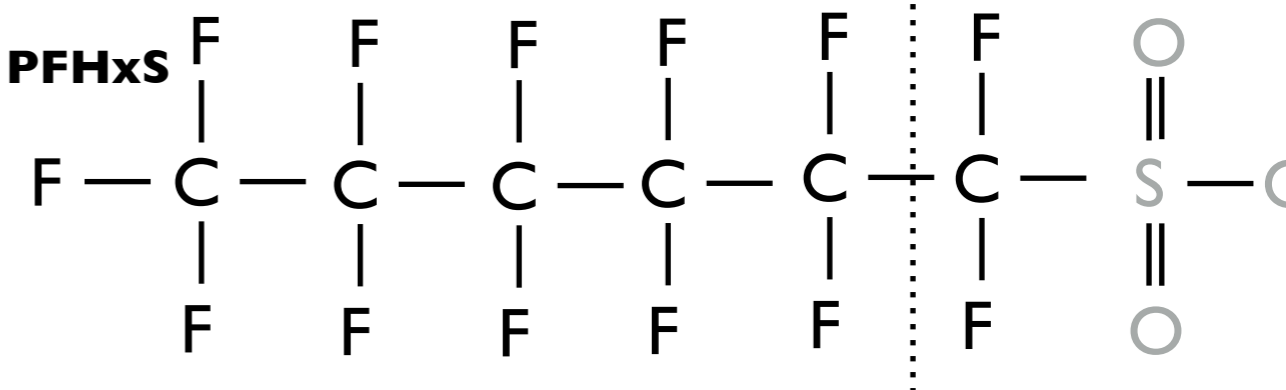
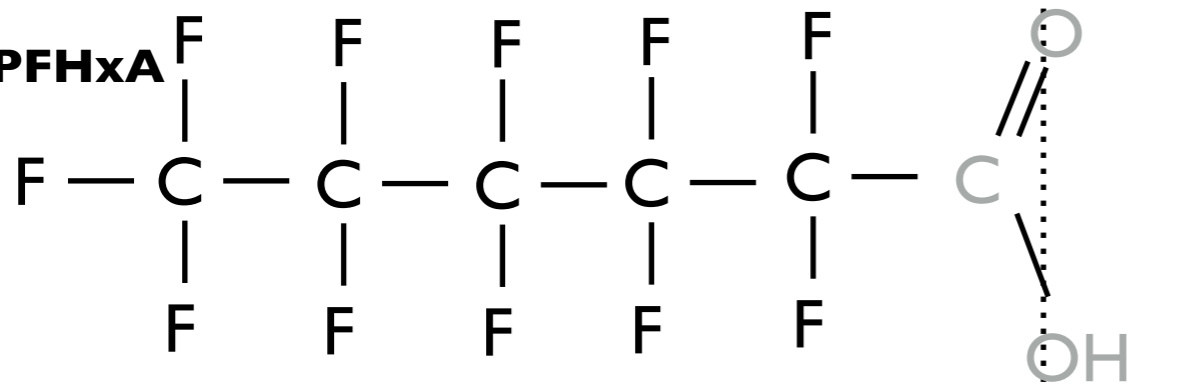
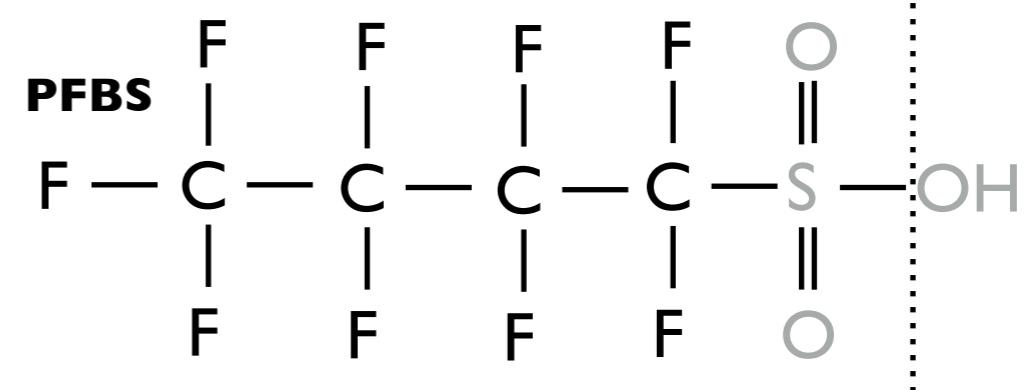
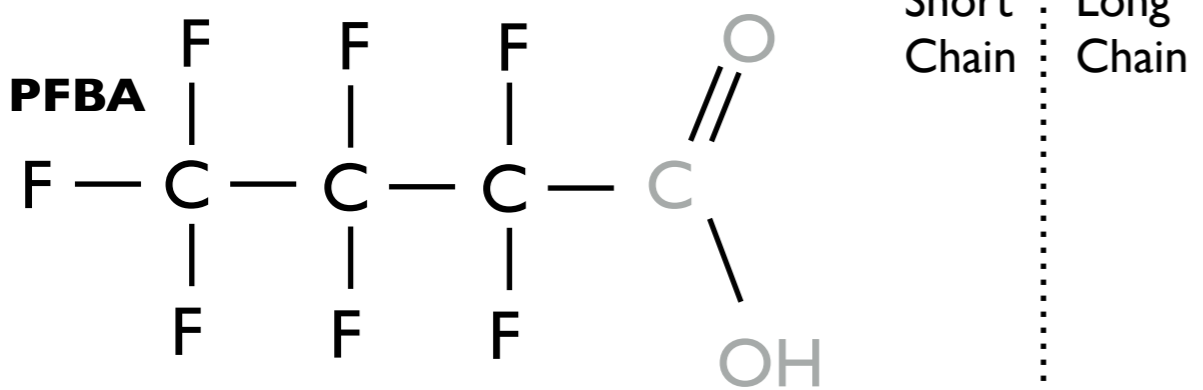
Perfluorinated carboxylate chemicals

Perfluoroalkyl sulfonate chemicals

Carboxylic Acids

Sulfonic Acids

Short Chain | Long Chain



Non-Polar
(Likes oils)

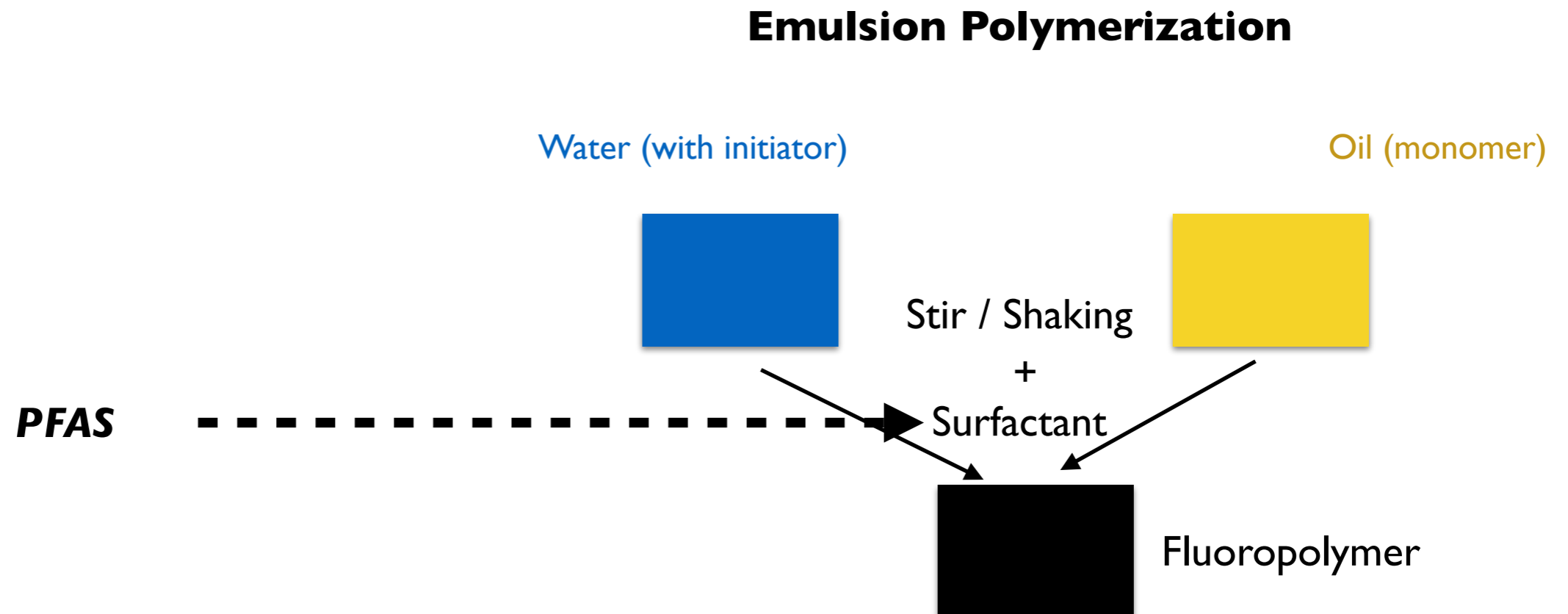
Non-Persistent | Persistent

Polar
(Likes water)

Non-Persistent | Persistent

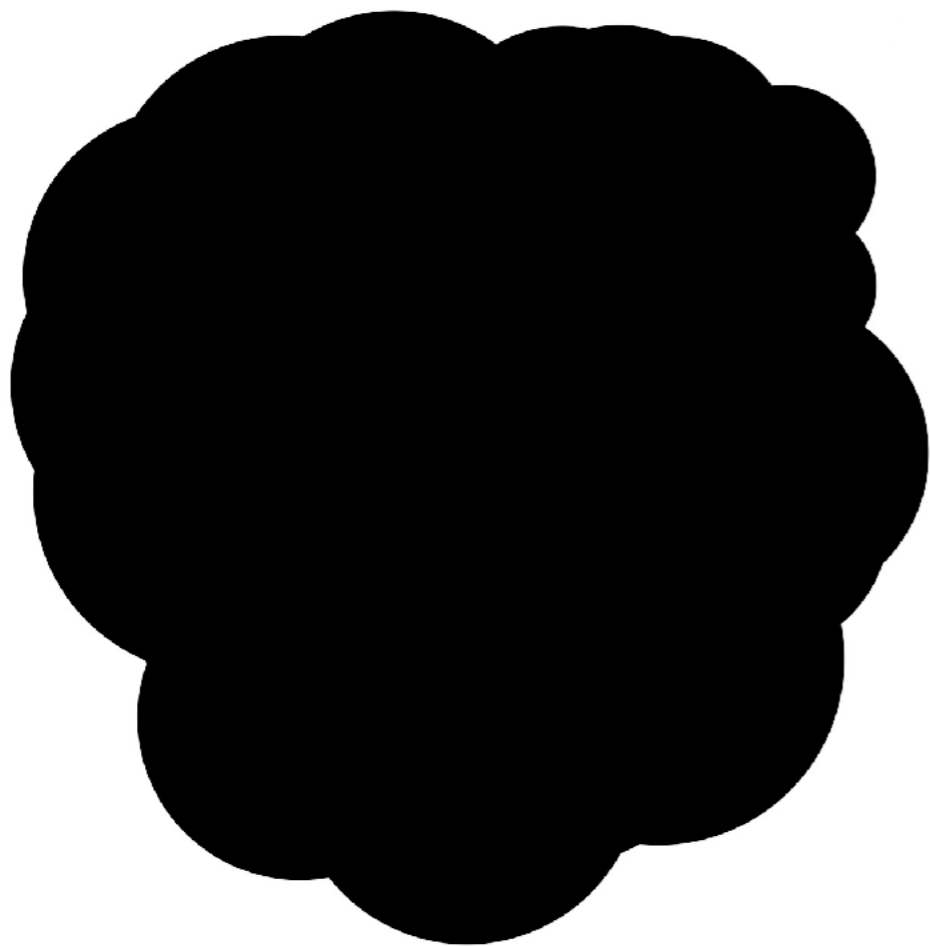
Emulsion Polymerization

- Emulsion
 - Oil and water mixture to polymerize
 - Needs surfactant to bind materials in oil to water
 - ie. LC-PFAC risk

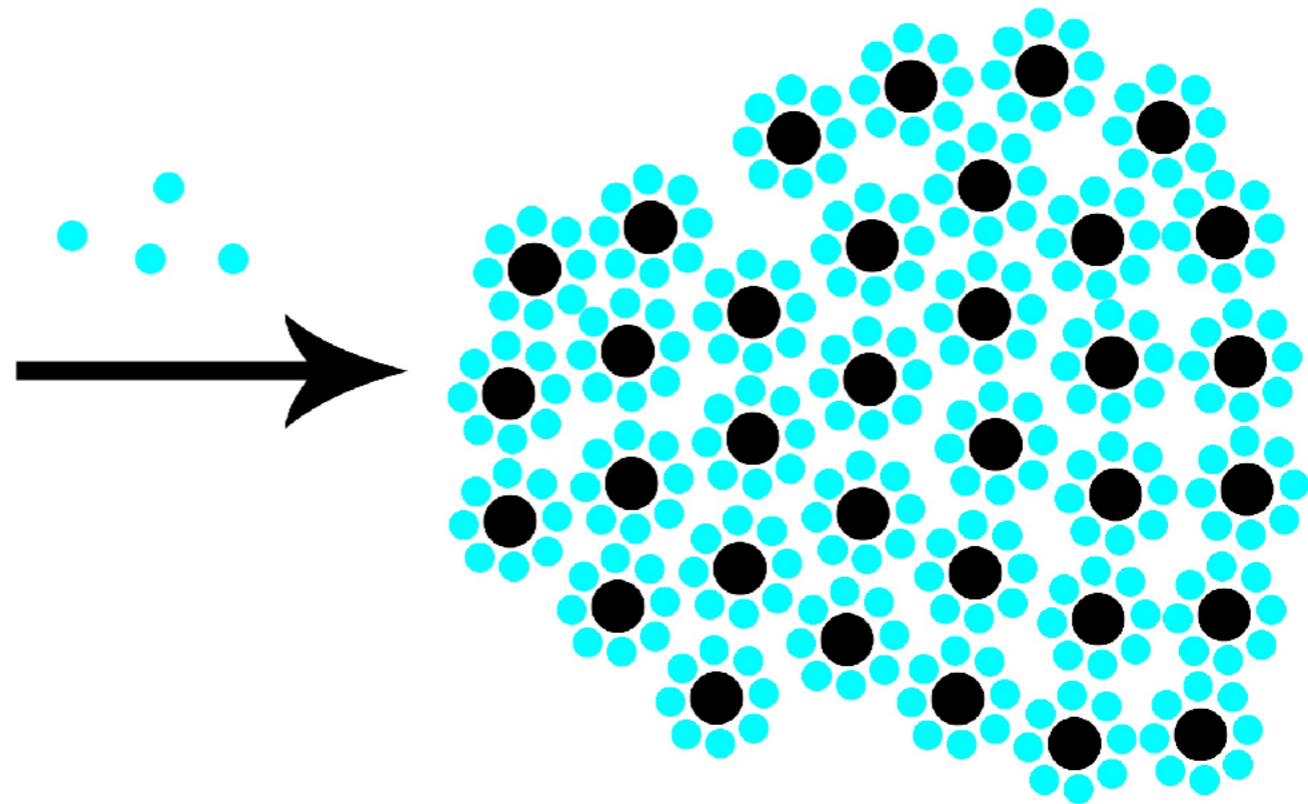


Surfactant

- Surfactant
 - Breaks up the larger oil (monomer) globules and makes them into more water friendly small globules
 - Allows more reactivity between oil and water



Oil (Monomer)



 = Surfactant

Polyfluoro uses Solid Polymer

- Solid fluoropolymer made from fluoropolymer

- PTFE connector
- PVDF tube
- FKM o-ring



- **PFOA risk** - as a surfactant in the manufacturing process

- Fluorination of another polymer

- Fluorine and oxygen gas process
- Example - fluorinated polyethylene

- **PFOA risk** - oxygenation generates carboxylates including PFOA and family

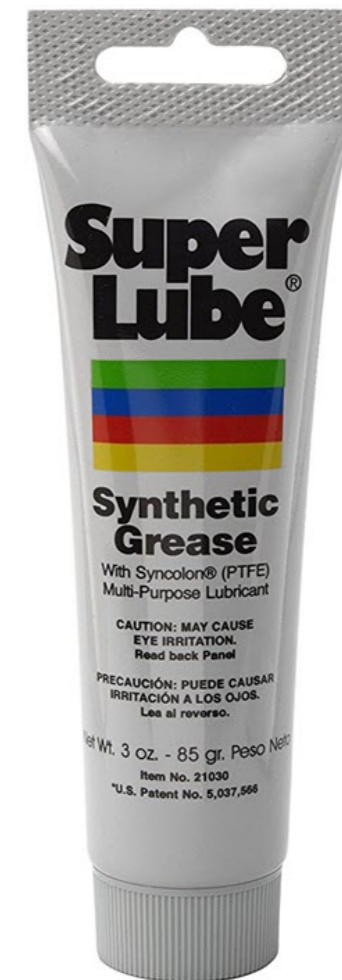


PFAS Coatings

- Polyfluoro cured on surface
 - PTFE water spray and high temperature curing into a solid
 - **PFOA risk** - as a surfactant in the powder manufacturing
- Polyfluoro suspended in another polymer
 - PTFE powder suspended in an acrylic or other polymer
 - **PFOA risk** - as a surfactant in the powder manufacturing
- Fluoroacrylate coating
 - Fluoroacrylate or sulphonate coating
 - **PFOA risk** - degradation of fluoroacrylate into PFOA

PFAS Lubricants

- Polyfluoro lubricants
 - PTFE powder suspending in mineral oil (or similar)
 - **PFOA risk** - as a surfactant in the powder manufacturing



Cosmetics

- Ingredient
 - C9-15 Fluoroalcohol Phosphate
- C9-15 Fluoroalcohol Phosphate
 - **PFOA risk** - degrades easily into PFOA
- Use
 - Higher spf make-up

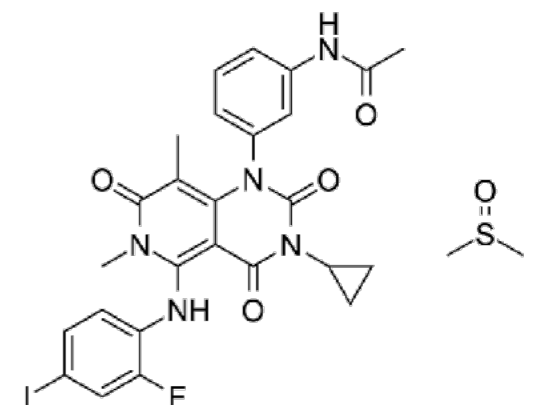


Miscellaneous

- Lots of miscellaneous uses
 - Fire extinguishing foam (Sulphonated fluorotelomers)
 - Lithium ion batteries (PVDF cathode binder)
 - Fluorinated medicines
 - (Which may be removed from reporting in definition updates)



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How to Comply Reporting vs Restriction

- **PFAS reporting**

- Presence of intentionally added PFAS
- Dominated by fluorine presence
- Virtually always fluoropolymers or fluoropolymer coatings



- **PFOA restriction**

- Restriction on manufacturing aids and degradation products of fluoropolymers
- Requires detailed testing of fluorinated substances



Typical questions we get

- What specific PFAS Chemicals are you able to test for?
- What test methods are available to test for PFAS?
- What types of samples can be used for these tests?

Starting Point Simplification

- For articles (physical products)
- Fluoropolymers (ie. PTFE)
 - Reportable
 - Non-restricted
 - Unless they contain restricted PFOA or similar.
- Non-Polymer PFAS (ie. PFOA)
 - Not generally reportable (not intentionally added)
 - Select groups are restriction (such as PFOA and longer)

Compliance Process

Measuring Fluorine Concentration

- Process for PFAS compliance
- **Step I**
 - Test directly for fluorine
 - No fluorine. No fluoroalkyl substances.
- WDXRF is the fast approach to identification of low levels of fluorine concentration



Compliance Process

Identification of Fluoro

- Process for PFAS compliance
- **Step 2**
 - Identification of intentionally added fluoroalkyl substance
 - Virtually all intentionally added PFAS are fluoropolymers or fluoropolymers suspended in another material
- Identification by
 - FTIR, supplemented by
 - Secondary elements (such as S and Si)



Compliance Process

Identification of Restricted Perfluoro

- Process for PFAS compliance
- **Step 3**
 - Measuring PFOA substances in fluoropolymers and fluoro coatings by LC-MS/MS
 - The major uses of PFOA (and similar) can be identified by polyfluoro present
- Identification and quantification by
 - LC-MS/MS



Claigan Approach

- PFAS compliance
 - **Step 1** - Fluorine screening
 - **Step 2** - Polymer identification
 - **Step 3** - (Restriction) - LC-MS/MS of the identified polymer



Typical questions we get

- What specific PFAS Chemicals are you able to test for?
 - **Step 1 - Test for fluorine**
 - **Step 2 - Identify fluoropolymer**
 - **Step 3 (restriction) - Test for PFOA and similar**
- What test methods are available to test for Fluorine?
 - **See above**
- What types of samples can be used for these tests? (Fabrics, plastics, etc?)
 - **All.**
 - Note - a common PFAS is a fluoro coating on metal filter screens

Claigan Approach

- PFAS compliance
 - **Step 1** - Fluorine screening
 - **Step 2** - Polymer identification
 - **Step 3** - (Restriction) - LC-MS/MS of the identified polymer



Q&A

